

# 1,2-Bis(3-thienyloxy)ethane: a thiophene-based precursor for thiophene-based azacryptand Mannich bases

Gaël Labat<sup>a</sup> and Joan Halfpenny<sup>b\*</sup><sup>a</sup>Institut de Chimie, Université de Neuchâtel, Avenue de Bellevaux 51, CH-2007 Neuchâtel, Switzerland, and <sup>b</sup>Department of Chemistry and Physics, Nottingham Trent University, Clifton Lane, Nottingham NG11 8NS, England

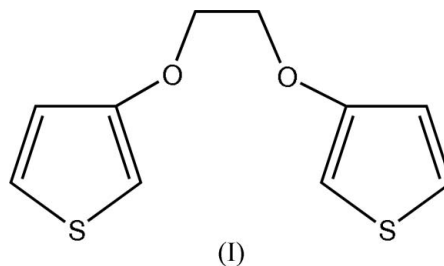
Correspondence e-mail: gael.labat@unine.ch

**Key indicators**Single-crystal X-ray study  
 $T = 153$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.031  
 $wR$  factor = 0.090  
Data-to-parameter ratio = 21.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_{10}\text{H}_{12}\text{O}_2\text{S}_2$ , is composed of two thiophene rings bridged by an  $-\text{O}(\text{CH}_2)_2\text{O}-$  chain in a *trans* arrangement. The molecule possesses  $C_2$  symmetry with the twofold axis bisecting the central C—C bond. In the crystal structure, molecules related by a centre of symmetry are bridged by  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds, forming a zigzag one-dimensional chain extending in the *c*-axis direction.

Received 26 July 2005  
Accepted 1 August 2005  
Online 6 August 2005**Comment**

The preparation of a range of open-chain cryptand-like structures, incorporating thiophene rings, as precursors for azacryptand Mannich bases, has been described by Barker *et al.* (1993) and Chaffin *et al.* (2001, 2002). The title compound, (I), was prepared by the reaction of methyl 3-hydroxythiophene-2-carboxylate with 1,2-dichloroethane and anhydrous potassium carbonate in anhydrous *N,N*-dimethylformamide, followed by saponification and decarboxylation.



The molecular structure of (I) is illustrated in Fig. 1, and selected bond distances and angles are given in Table 1. In compound (I), two thiophene rings are bridged by an  $-\text{O}(\text{CH}_2)_2\text{O}-$  chain in a *trans* arrangement. A twofold axis bisects the central ethane bond [ $\text{C}5-\text{C}5(1-x, y, \frac{1}{2}-z)$ ] and each half of the molecule is almost planar, with  $\text{C}5-\text{O}1-\text{C}2-\text{C}1$  and  $\text{C}5-\text{O}1-\text{C}2-\text{C}3$  torsion angles of  $0.00$  (18) and  $-178.45$  (11) $^\circ$ , respectively. The bond lengths and angles (Table 1) are similar to those in an unsubstituted thiophene described by Bonham & Momany (1963).

The crystal packing of compound (I) is illustrated in Fig. 2. The molecules related by centres of symmetry are linked by  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds; details are given in Table 2. It can be seen that the molecules are arranged in a such a way as to form a zigzag one-dimensional polymer extending in the crystallographic *c*-axis direction.

**Experimental**

Compound (I) was synthesized according to the procedure described by Chaffin *et al.* (2001). Suitable crystals for X-ray crystallography

analysis were obtained by slow evaporation of a 1:1 ethanol-dichloromethane solution.

Crystal data

C<sub>10</sub>H<sub>10</sub>O<sub>2</sub>S<sub>2</sub>  
*M<sub>r</sub>* = 226.30  
 Monoclinic, *C*2/*c*  
*a* = 22.175 (3) Å  
*b* = 5.3918 (4) Å  
*c* = 9.0831 (11) Å  
 β = 111.362 (9)°  
*V* = 1011.39 (19) Å<sup>3</sup>  
*Z* = 4

*D<sub>x</sub>* = 1.486 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 Cell parameters from 9090 reflections  
 θ = 1.9–29.6°  
 μ = 0.49 mm<sup>-1</sup>  
*T* = 153 (2) K  
 Plate, colourless  
 0.5 × 0.5 × 0.2 mm

Data collection

Stoe IPDS-II diffractometer  
 ω scans  
 Absorption correction: none  
 9441 measured reflections  
 1398 independent reflections  
 1302 reflections with *I* > 2σ(*I*)

*R*<sub>int</sub> = 0.053  
 θ<sub>max</sub> = 29.4°  
*h* = -30 → 30  
*k* = -7 → 7  
*l* = -11 → 12

Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.031  
*wR* (*F*<sup>2</sup>) = 0.090  
*S* = 1.03  
 1398 reflections  
 64 parameters  
 H-atom parameters constrained

*w* = 1/[σ<sup>2</sup>(*F*<sub>o</sub><sup>2</sup>) + (0.0472*P*)<sup>2</sup> + 1.0038*P*]  
 where *P* = (*F*<sub>o</sub><sup>2</sup> + 2*F*<sub>c</sub><sup>2</sup>)/3  
 (Δ/σ)<sub>max</sub> = 0.001  
 Δρ<sub>max</sub> = 0.35 e Å<sup>-3</sup>  
 Δρ<sub>min</sub> = -0.35 e Å<sup>-3</sup>

Table 1

Selected geometric parameters (Å, °).

S1–C4	1.7129 (14)	C3–C4	1.3674 (19)
S1–C1	1.7178 (13)	C3–C2	1.4227 (17)
O1–C2	1.3597 (15)	C2–C1	1.3672 (17)
O1–C5	1.4288 (15)	C5–C5 <sup>i</sup>	1.500 (2)
C4–S1–C1	92.55 (6)	O1–C2–C3	118.83 (11)
C2–O1–C5	115.12 (10)	C3–C4–S1	111.56 (10)
C4–C3–C2	111.88 (11)	O1–C5–C5 <sup>i</sup>	108.13 (9)
C1–C2–O1	127.57 (12)	C2–C1–S1	110.43 (10)
C1–C2–C3	113.59 (12)		

Symmetry code: (i) -*x* + 1, *y*, -*z* + ½.

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
C3–H3...O1 <sup>ii</sup>	1.00	2.41	3.3940 (16)	170

Symmetry code: (ii) -*x* + 1, -*y*, -*z*.

H atoms were located in difference Fourier maps and held fixed with *U*<sub>iso</sub>(H) = 0.05 Å<sup>2</sup> and C–H = 0.94–1.05 Å.

Data collection: *X*-AREA (Stoe & Cie, 2002); cell refinement: *X*-AREA; data reduction: *X*-RED32 (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

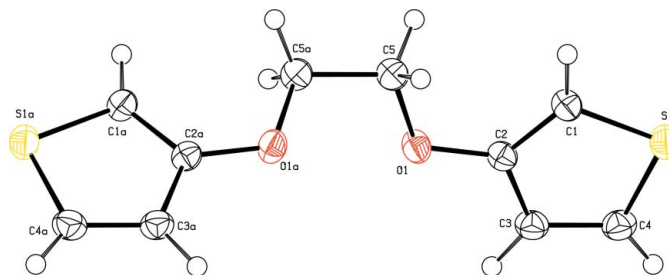


Figure 1

View of compound (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry code: (a) 1 - *x*, *y*, ½ - *z*.]

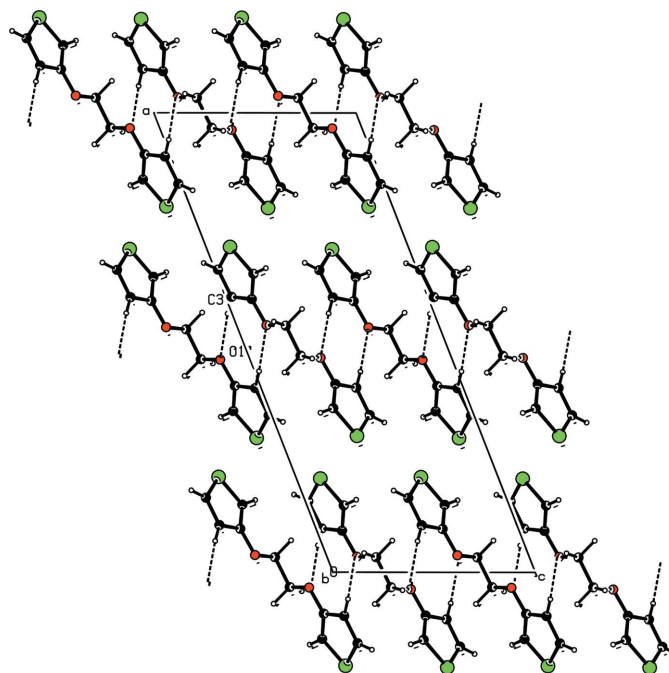


Figure 2

The crystal packing of compound (I), viewed down the *b* axis. C–H...O hydrogen bonds are shown as dashed lines (details are given in Table 2).

The authors thank Professor Helen Stoeckli-Evans (Université de Neuchâtel) for making available the Stoe *IPDS* diffractometer for data collection.

References

- Barker, J. M., Chaffin, J. D. E., Halfpenny, J., Huddleston, P. R. & Tseki, P. F. (1993). *J. Chem. Soc. Chem. Commun.* pp. 1733–1734.  
 Bonham, R. A. & Momany, F. A. (1963). *J. Phys. Chem.* **67**, 2474–2477.  
 Chaffin, J. D. E., Barker, J. M. & Huddleston, P. R. (2001). *J. Chem. Soc. Perkin Trans. 1*, pp. 1398–1405.  
 Chaffin, J. D. E., Barker, J. M. & Huddleston, P. R. (2002). *J. Chem. Soc. Perkin Trans. 1*, pp. 717–724.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.  
 Stoe & Cie (2002). *X*-AREA (Version 1.17) and *X*-RED32 (Version 1.04). Stoe & Cie GmbH, Darmstadt, Germany.